A method of field sample collection for the detection of arsenic species As(III), As(V) monomethylarsinate (MMA), and dimethylarsinate (DMA) in drinking water which comprises:

- 5
- (a) providing three separate solid-phase extraction columns for a serial extraction of the drinking water in the field wherein
- 10
- (1) a first of the columns which is a cation exchange column comprising a first medium for ion exchanging which removes interfering cation ions in the water from the As(V), MMA, DMA, and As(III) in the water;
- 15
- exchange column comprising a second medium at a pH of about 1.0 to 1.5 for ion exchanging which removes the DMA in the water from the As(V), MMA, and As(III) in the water; and

(2) a second of the columns which is a cation

a third of the columns which is

comprising a

dioctyltin

- 20
- dichloride coated medium at a pH of 2.5 to 3.5 which removes the As(V) and MMA in the water while the As(III) remains in the water; and

  (b) serially introducing the water into the

first, second, and third columns, and collecting a first

effluent from the third column containing the As(III).

column

(3)

exchange/sorption

-2-

The method of Claim 1 wherein the first medium for the first of the columns is a weak acid cation exchange media.

The method of Claim 2 wherein the weak acid cation medium comprises a functional group selected from the group consisting of iminodiacetate, amino phosphonic, phosphinic, and methylene thiol.

-4-

The method of Claim 3 wherein the functional group is iminodiacetate.

-5-

The method of Claim 2, 3, or 4 wherein the first medium comprises polystyrene divinylbenzene.

-6-

The method of Claim 1 wherein the second medium for the second of the columns is a strong acid cation exchange medium.

-7-

The method of Claim 6 wherein the strong acid cation medium comprises a sulphonate functional group.

-8-

The method of Claim 6 or 7 wherein the second medium comprises polystyrene divinylbenzene.

-9-

The method of Claim 1 wherein the third medium comprises calcined diatomaceous earth.

A kit for collection of a field sample for the detection of arsenic species As(III), As(V), monomethylarsinate (MMA), and dimethylarsinate (DMA) in drinking water which comprises:

three separate solid-phase extraction columns

wherein

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(1) a first of the columns which is a cation exchange column comprising a first medium for ion exchanging which removes interfering cation ions in the water from the As(V), MMA, DMA, and As(III) in the water;

- (2) a second of the columns which is a cation exchange column comprising a second medium at a pH of about 1.0 to 1.5 for ion exchanging which removes the DMA in the water from the As(V), MMA, and As(III) in the water; and
- (3) a third of the columns which is an exchange/sorption column comprising a dioctyltin dichloride coated third medium at a pH of 2.5 to 3.5 which removes the As(V) and MMA in the water while the As(III) remains in the water.

-11-

The kit of Claim 10 wherein the first medium of the first of the columns is a weak acid cation exchumnge media.

**-**12-

The kit of Claim 11 wherein the weak acid cation medium comprises a functional group selected from the group consisting of iminodiacetate, amino phosphonic, phosphinic, and methylene thiol immobilized thereon.

The kit of Claim 12 wherein the functional group is iminodiacetate.

-14-

The kit of Claim 11, 12, or 13 wherein the first medium comprises polystyrene divinylbenzene.

-15-

The kit of Claim 10 wherein the second medium of the second of the columns is a strong acid cation exchange media.

-16-

The kit of Claim 15 wherein the strong acid cation medium comprises a sulphonate functional group.

-17-

The kit of Claim 15 or 16 wherein the second medium comprises polystyrene divinylbenzene.

-18-

The kit of Claim 10 wherein the third medium of the third column comprises calcined diatomaceous earth.

A method of field sample collection for the detection of arsenic species As(III), As(V) monomethylarsinate (MMA), and dimethylarsinate (DMA) in drinking water which comprises:

- (a) providing three separate solid-phase extraction media for a serial extraction of the drinking water in the field wherein
- (1) a first of the media removes interfering cation ions in the water from the As(V), MMA, DMA, and As(III) in the water;
- (2) a second of the media at a pH of about 1.0 to 1.5 removes the DMA in the water from the As(V), MMA, and As(III) in the water; and
- (3) a third of the media at a pH of 2.5 to 3.5 removes the As(V) and MMA in the water while the As(III) remains in the water; and
- (b) serially introducing the water into the first, second, and third media, and collecting a first effluent from the third media containing the As(III).

-20-

The method of Claim 19 wherein the first of the media is a weak acid cation exchange medium.

-21-

The method of Claim 20 wherein the weak acid cation medium comprises a functional group selected from the group consisting of iminodiacetate, amino phosphonic, phosphinic, and methylene thiol.

-22-

The method of Claim 21 wherein the functional group is iminodiacetate.

-23-

The method of Claim 20, 21, or 22 wherein the first of the media comprises polystyrene divinylbenzene.

-24-

The method of Claim 19 wherein the second of the media is a strong acid cation exchange medium.

-25-

The method of Claim 24 wherein the strong acid cation medium comprises a sulphonate functional group.

-26-

The method of Claim 24 or 25 wherein the second of the media comprises polystyrene divinylbenzene.

-27-

The method of Claim 19 wherein the third of the media comprises calcined diatomaceous earth.

A kit for collection of a field sample for the detection of arsenic species As(III), As(V), monomethylarsonate (MMA), and dimethylarsinate (DMA) in drinking water which comprises:

three separate solid-phase extraction media wherein

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- (1) a first of the media removes interfering cation ions in the water from the As(V), MMA, DMA, and As(III) in the water;
- 10 (2) a second of the media which at a pH of about 1.0 to 1.5 removes the DMA in the water from the As(V), MMA, and As(III) in the water; and
  - (3) a third of the media which at a pH of 2.5 to 3.5 removes the As(V) and MMA in the water while the As(III) remains in the water.

## -29-

The kit of Claim 28 wherein the first of the media is a weak acid cation exchange medium.

## -30-

The kit of Claim 29 wherein the weak acid cation medium comprises a functional group selected from the group consisting of iminodiacetate, amino phosphonic, phosphinic, and methylene thiol immobilized thereon.

## -31-

The kit of Claim 30 wherein the functional group is iminodiacetate.

## -32-

The kit of Claim 29, 30, or 31 wherein the first of the media comprises polystyrene divinylbenzene.

The kit of Claim 28 wherein the second of the media is a strong acid cation exchange medium.

-34-

The kit of Claim 15 wherein the strong acid cation medium comprises a sulphonate functional group.

-35-

The kit of Claim 33 or 34 wherein the second of the media comprises polystyrene divinylbenzene.

-36-

The kit of Claim 28 wherein the third of the media comprises dioctyltin dichlrodie coated calcined diatomaceous earth.